

Acid hydrolysates of a variety of bovine trypsin preparations have been investigated. Twice-recrystallized trypsin from Worthington Biochemicals and from Mann Chemicals, a five-times recrystallized preparation from Worthington Biochemicals, a three-times recrystallized preparation made in our laboratory and a three-times recrystallized preparation of diisopropylphosphoryl-trypsin were included in these studies. The presence of hydroxylysine, demonstrated in hydrolysates of all of these preparations, suggests that this amino acid is a constituent of the trypsin molecule.

In a previous amino acid analysis of trypsin reported from this laboratory¹¹ the hydroxylysine was undetected, apparently because the basic amino acids were determined by use of the 15-cm column² for basic amino acids instead of the 50-cm column³ used in this work.

In one analysis of an acid hydrolysate of bovine chymotrypsin a total of 0.6 mole of hydroxylysine and allohydroxylysine per mole of protein was found.

The details of this work will be published elsewhere.

National Institute of Arthritis and Metabolic Diseases,
National Institutes of Health, Bethesda, Maryland (U.S.A.)

T. VISWANATHA
F. IRREVERRE

- ¹ G. H. DIXON, D. L. KAUFFMAN AND H. NEURATH, *J. Am. Chem. Soc.*, 80 (1958) 1260.
- ² T. VISWANATHA, *Compt. rend. trav. lab. Carlsberg, Sér. chim.*, 30 (1957) 13, 183.
- ³ D. H. SPACKMAN, W. H. STEIN AND S. MOORE, *Anal. Chem.*, 30 (1958) 1190.
- ⁴ A. V. GUNTELBERG AND M. OTTESEN, *Compt. rend. trav. lab., Carlsberg, Sér. Chim.*, 29 (1954) 36.
- ⁵ J. A. COHEN, R. A. OOSTERBAAN, H. S. JANEZ AND F. BERENDS, in *Symposium on Enzyme Reaction Mechanisms*, Oak Ridge National Laboratory, Oak Ridge, Tennessee, 1959.
- ⁶ J. E. FOLK, J. A. GLADNER AND T. VISWANATHA, *Biochim. Biophys. Acta*, 36 (1959) 256.
- ⁷ P. B. HAMILTON AND R. A. ANDERSON, in A. B. MEISTER, *Biochemical Preparations*, in the press.
- ⁸ D. P. SCHWARTZ, *Anal. Chem.*, 30 (1958) 1855.
- ⁹ F. IRREVERRE AND W. MARTIN, *Anal. Chem.*, 26 (1954) 257.
- ¹⁰ K. A. PIEZ, F. IRREVERRE AND H. L. WOLFF, *J. Biol. Chem.*, 223 (1956) 687.
- ¹¹ T. VISWANATHA, W. B. LAWSON AND B. WITKOP, *Biochim. Biophys. Acta*, in the press.

Received March 28th, 1960

Biochim. Biophys. Acta, 40 (1960) 564-565

A nucleotide enzyme complex associated with fowl leukemia virus

The virus particles that can be isolated from blood plasma in the case of fowl myeloblastic leukemia possess a high magnesium-stimulated ATPase activity¹⁻³. ADP has been claimed to act as a competitive inhibitor in the hydrolysis of ATP⁴.

A closer study of the viral ATPase activity reveals properties, some of which indicate a complex of phosphorylating enzymes. If activated by Mg^{++} , two pH optima can be distinguished (pH 7.0 and 8.5). Besides Mg^{++} , also other metal ions activate

Abbreviations: NTP, ATP, UTP, CTP and GTP for the 5'-triphosphates; NDP, ADP, UDP, CDP and GDP for the 5'-diphosphates; NMP, AMP, UMP, CMP and GMP for the 5'-monophosphates of nucleosides, adenosine, uridine, cytosine and guanosine; P_i , inorganic orthophosphate; Tris, tris(hydroxymethyl)aminomethane; EDTA, ethylenediaminetetraacetic acid.

Biochim. Biophys. Acta, 40 (1960) 565-567

the enzyme complex in the proportions: Mg^{++} 1.0, Cd^{++} 0.5, Fe^{++} 0.5, Zn^{++} 0.25 and Cu^{++} 0.25.

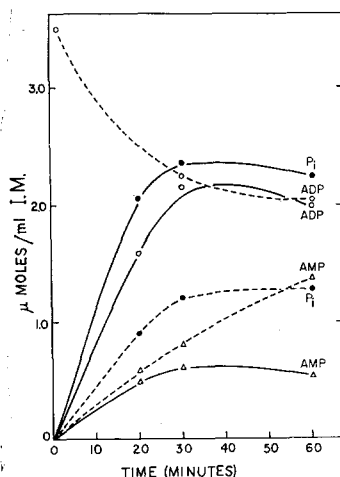
The enzyme complex splits inorganic orthophosphate from nucleoside diphosphates at pH 7.0 in the sequence of activities: $ADP < UDP < CDP < GDP$, and at pH 8.5 $UDP < ADP = CDP < GDP$. The quantitative analysis of these reactions shows that at least two events take place: $NDP \rightleftharpoons NMP + P_1$ (ref. 5, 6); and $n\text{ }NDP \rightleftharpoons (NMP)_n + n\text{ }P_1$ (ref. 8). To these reactions can be added the ATPase activity: $ATP \rightarrow ADP + P_1$, when ATP is used as the substrate.

Thus, in addition to a high triphosphatase activity, a polynucleotide-phosphorylase (polyase) activity is present, which reaction runs most efficiently if GDP is used as the substrate ($GDP \gg CDP > UDP > ADP$, see Table I).

TABLE I

Enzyme reactions with fowl leukemia virus isolated from blood plasma^{2,14}. Viral enzyme concentrations, 2.0 and 8.0 $\mu\text{g N/ml}$ reaction mixture¹⁵. Incubation medium: 0.25 M sucrose and 0.04 M Tris-buffer. P_1 estimation according to ref. 16. The analyses of the polyase, nucleoside diphosphatase and myokinase activities were made in principle as described by BEERS¹⁷. Chromatography of the reaction mixtures was performed on Whatman paper with the isobutyric acid- NH_3 -EDTA mixture as solvent. The spots were eluted with 0.1 N NH_4OH for spectrophotometric determinations.

Substrate	$\mu\text{moles } P_i/\text{mg N}/30\text{ min}$ Substrate, 2.5 mM; Mg^{++} , 3.5 mM; sucrose-Tris; 37°		(NMP) $_n$ as $\mu\text{moles NMP}/\text{mg N}$ in 30 min	
	pH 7.0	pH 8.5	Substrate, 6.1 mM; Mg^{++} , 1.5 mM; Tris, pH 8.5; 37°	
ADP	36.6	40.6	4.61	38.0
UDP	40.0	33.3	7.32	29.4
CDP	48.4	41.3	11.95	40.0
GDP	58.9	53.9	65.0	58.0
ATP	364.0	408.0	> ADP	445.0



The synthesizing polyase activities are associated with a diphosphatase and possibly also a myokinase-like activity according to the terminology of GIBSON *et al.*⁷. This conclusion is drawn from the decrease of AMP during the later part of the incubation (> 30 min, ATP as substrate, see Fig. 1), as well as from the simultaneous decrease of P_1 in the same period. The sequence of activities when different nucleoside diphosphates are incubated is similar to that of the polyase: $GDP > CDP > ADP > UDP$.

Fig. 1. Viral enzyme complex incubated with 2.5 mM ATP (—) or 4.0 mM ADP (---). Incubation medium (I.M.): 0.25 M sucrose; 0.04 M Tris, pH 8.5; 1.5 mM Mg^{++} ; 37°. ADP \circ , AMP Δ , P_1 \bullet .

If the activities of the enzyme complex are expressed in specific activities (units/mg N/min), the figures will roughly correspond to the specific activities measured in

purified enzymes from other sources, *e.g.* the adenylate kinase, nucleoside diphosphatase from liver^{6,7}. The six to ten times higher activity of the GMP phosphorylase in comparison with the other polyase activities is noteworthy from the point of view of specificity and heterogeneity⁸⁻¹⁰ of the polyase enzymes and also of the tumor polynucleotides¹¹.

Finally, it should be mentioned that ATPase activity with characteristics of the electron-transport chain is also present. Perhaps the viral enzyme complex represents a reminiscence of those parts in the host cell which have participated in the virus synthesis¹². The view¹³ "that viruses do not act only as cell and tissue destroyers but as parts of normal animal cells" conforms with this hypothesis.

This work was supported by a grant from the Swedish Cancer Society.

Department of Pathology, Karolinska Institute, Stockholm (Sweden)

J. RÍMAN*

B. THORELL

- ¹ E. E. MOMMAERTS, D. G. SHARP, E. A. ECKERT, D. BEARD AND J. W. BEARD, *J. Natl. Cancer Inst.*, 14 (1954) 1011.
- ² I. GREEN AND D. G. SHARP, *Biochim. Biophys. Acta*, 18 (1955) 36.
- ³ J. W. BEARD, *American Scientist*, 46 (1958) 226.
- ⁴ I. GREEN, *Biochim. Biophys. Acta*, 18 (1955) 43.
- ⁵ J. D. GREGORY, *Fed. Proc.*, 14 (1955) 221.
- ⁶ G. W. E. PLAUT, *J. Biol. Chem.*, 217 (1955) 235.
- ⁷ D. M. GIBSON, P. AYENGAR AND D. R. SANADI, *Biochim. Biophys. Acta*, 16 (1955) 536.
- ⁸ M. GRUNBERG-MANAGO, P. J. ORTIZ AND S. OCHOA, *Biochim. Biophys. Acta*, 20 (1956) 269.
- ⁹ P. S. OLMSTED, *Biochim. Biophys. Acta*, 27 (1958) 222.
- ¹⁰ A. A. HAKIZ, *Arch. Biochem. Biophys.*, 83 (1959) 390.
- ¹¹ G. DE LAMIRANTE, C. ALLARD AND A. CANTERO, *Cancer Research*, 15 (1955) 329.
- ¹² B. THORELL AND E. YAMADA, *Biochim. Biophys. Acta*, 31 (1959) 104.
- ¹³ H. KOPROWSKI, *Trans. N.Y. Acad. Sci.*, 22 (1960) 176.
- ¹⁴ G. S. BEAUDREAU AND C. BECKER, *J. Natl. Cancer Inst.*, 20 (1958) 339.
- ¹⁵ E. J. CONWAY AND E. O'MALLEY, *Biochem. J.*, 36 (1942) 43.
- ¹⁶ T. TEORELL AND B. NORBERG, *Biochem. Z.*, 249 (1932) 53.
- ¹⁷ R. BEERS JR., *Biochem. J.*, 66 (1957) 686.

Received March 25th, 1960

* Research Fellow of the World Health Organisation. Present address: Institute of Chemistry, Czechoslovak Academy of Science, Prague (Czechoslovakia).

Biochim. Biophys. Acta, 40 (1960) 565-567

Interference with adenine and histidine metabolism of microorganisms by aminotriazole

3-Amino-1,2,4-triazole (aminotriazole), a heterocyclic nitrogen compound used extensively in agriculture as a weed killer, bears a close structural relationship to 4-aminoimidazole. 4-Aminoimidazole is recognized as an intermediate in purine degradation in *Clostridium cylindrosporum*¹ and 4-aminoimidazole ribonucleotide is an intermediate in purine biosynthesis². From these considerations, it might be

Biochim. Biophys. Acta, 40 (1960) 567-569